

Aminoacetyl Moiety as a Potential Surrogate for Diacylhydrazine Group of SC-51089, a Potent PGE₂ Antagonist, and Its Analogs

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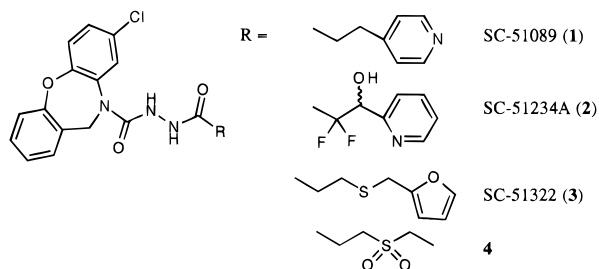
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8-Chlorodibenz[*b,f*][1,4]oxazepine-10(11*H*)-carboxylic acid, 2-[1-oxo-3-(4-pyridinyl)propyl]hydrazide, monohydrochloride (**1**, SC-51089) is a functional PGE₂ antagonist selective for the EP₁ receptor subtype with antinociceptive activity. During metabolism in cultured rat hepatocytes, SC-51089, which contains a diacylhydrazine moiety, has been shown to release hydrazine. Analogs of SC-51089, in which the diacylhydrazine functionality has been replaced by isosteric and isoelectronic groups, have been synthesized and have been shown to be analgesics and PGE₂ antagonists of the EP₁ subtype. This report discusses the structure–activity relationships within these series.

Introduction

SC-51089, a functional PGE₂ antagonist with an excellent analgesic profile,^{1,2} contains a diacylhydrazine moiety, which, during metabolism in cultured rat hepatocytes, releases hydrazine.³ Although hydrazine was known to be carcinogenic in rodents,⁴ its release had not been seen in an earlier member of this structural class, **4**.³ This phenomenon necessitates identifying a pharmacophore that can replace the diacylhydrazine and, yet, will retain the desirable pharmacological profile of SC-51089.



Pain on exposure to noxious stimuli is thought to occur through the intermediacy of several autacoids, among them prostaglandin E₂ (PGE₂), that are released at the site of injury.⁵ Relief of mild to moderate pain is attained with over-the-counter and prescription non-steroidal antiinflammatory drugs (NSAIDs). The antinociceptive action of NSAIDs is believed to occur through the inhibition of the action of cyclooxygenase on arachidonic acid.⁶ Ferreira suggested that the effectiveness of NSAIDs against the pain accompanying inflammation and tissue injury is due specifically to the inhibition of PGE₂ formation from arachidonate.^{6b} More particularly, NSAIDs block the production of PGE₂, a prostanoid, which causes pain and hyperalgesia and potentiates the action of bradykinin in pain transmission.⁶ Two decades of research by several research groups have provided evidence supporting this view.⁷

While NSAIDs are considered generally safe, untoward effects do accompany their administration. The most notable and most prevalent are the gastric side

effects including dyspepsia and heartburn.⁸ Chronic use of NSAIDs often results in severe damage to gastric and intestinal mucosa that can lead to frank bleeding requiring emergency intervention.

The rationale of our analgesia program is based on the hypothesis that PGE₂-induced hyperalgesia occurring in inflamed tissue would be attenuated by selective blockade of PGE₂ receptors.^{6,7} Analgesia based on PGE₂ antagonism would preclude the problems associated with inhibition of prostanoid biosynthesis.

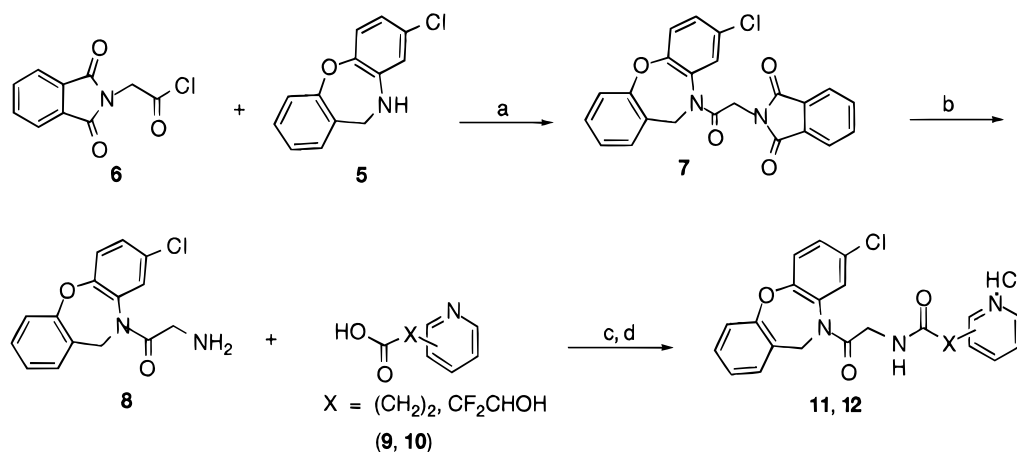
Although research on the SC-51089 class of compounds has exhaustively explored the extended chain^{1,9,10} and more recently the 8-chlorodibenzoxazepine,¹¹ little effort has been directed toward identifying an alternative for the diacylhydrazine functionality. Our initial strategy examined the need for hydrogen-bonding properties of the diacylhydrazine group. This was accomplished by a systematic replacement of each nitrogen of hydrazine with a methylene. Although conceptually basic, this is key to understanding the requirement for the hydrogen bond acceptor–hydrogen bond donor properties of the diacylhydrazine. Using the extended chains of four exceedingly effective analgesic-PGE₂ antagonists,^{1,10,12} the diacylhydrazine group of compounds **1–4** has been substituted by an aminoacetyl moiety as illustrated by analogs **11–13**, **15**, **21**, and **22**.

Synthesis

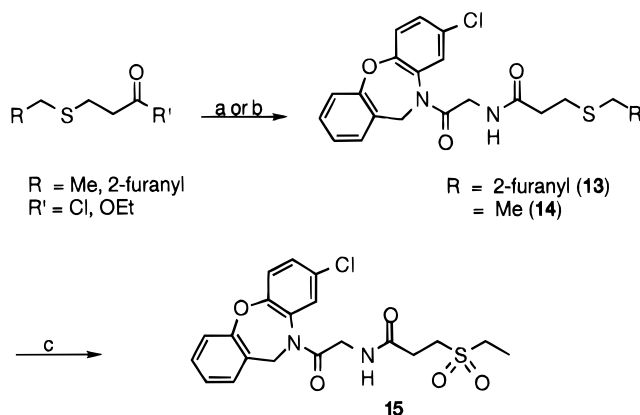
Syntheses of the diamide analogs, where the nitrogen proximal to the 8-chlorodibenzoxazepine ring is replaced with a methylene, are illustrated in Schemes 1 and 2. Acylation of 8-chlorodibenzoxazepine (**5**) with activated Boc-Gly (e.g. symmetrical anhydride, mixed anhydride, and carbodiimide with and without additives) was unsuccessful. With Phth-Gly-Cl, acylation of **5** was effected, yielding **7**. Subsequent transformations of **7** including removal of the phthaloyl protecting group and acylation of the resultant amine gave analogs **11–14**. Oxidation of **14** with H₂O₂ resulted in sulfone **15**.

The methods for assembling the urea–ketone analogs, in which the nitrogen distal to 8-chlorodibenzoxazepine is replaced with methylene, are illustrated in Scheme 3. Installation of the aminoacetyl group was accomplished by acylating **5** with ethyl isocyanoacetate.

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Scheme 1^a

^a (a) toluene, TEA, Δ , 3 h; (b) NH₂NH₂, EtOH, Δ , 4 h; (c) Me₂N(CH₂)₃N=C=NCH₂CH₃, Et₃N, DMF, 5–20 °C, 16 h; (d) HCl/EtOH or 1 N HCl.

Scheme 2^a

^a (a) **8**, AlMe₃, toluene; (b) **8**, TEA, DCM, 5 °C; (c) 30% H₂O₂, HOAc, 72 °C.

Subsequent manipulations, including saponication of **16** and acylation of *N,O*-dimethylhydroxylamine with **17**, resulted in the Weinreb amide **18**. Metalation of **18** with vinylmagnesium bromide yielded α,β -unsaturated ketone, **19**. Michael addition to **19** with ethyl mercaptan and furfuryl mercaptan produced **20** and **21**. Oxidation of **20** with H₂O₂ resulted in the sulfone **22**.

Pharmacology

The phenylbenzoquinone (PBQ) writhing assay in mouse was used to evaluate the antinociceptive effectiveness of these compounds.¹ PGE₂ antagonism was confirmed in PGE₂-stimulated guinea pig ileum muscle strip assay.¹

Results and Discussion

As seen in Table 1, the urea-ketones, **21** and **22**, show a reduction in analgesic efficacy as compared to parents, **3** and **4**, respectively. A loss in PGE₂ antagonism activity is also seen with **21**. Given the uninteresting activity in these urea-ketones, no attempts were made to synthesize the urea-ketone analogs of **1** and **2**.

A disparate picture emerges with the diamide congeners of the PGE₂ antagonists-analgesics as can be seen in Table 1. Diamide **15** has a biological profile that parallels its parent, **4**, in that it has comparable analgesic and PGE₂ antagonism activity. Analog **13**

disappointingly is a significantly less efficacious analgesic and PGE₂ antagonist than its parent, **3**.¹⁰ As with **13**, uninteresting analgesic and PGE₂ antagonism activity is noted with **11**. None of the desirable properties of its parent, **1**, is seen with **11**. Analog **12** parallels its parent, **2**, in its analgesic potency.¹

Conclusion

PGE₂ antagonism is observed in all compounds of this series that have been tested; however, this series of compounds shows a trend toward attenuation of PGE₂ antagonism. That PGE₂ antagonism is observed suggests that the hydrogen-bonding properties of the hydrazine NH's are not essential in eliciting PGE₂ antagonism in this class of compounds.

Observing the desired analgesic activity in diamides **12** and **15** indicates that modification of the diacylhydrazine is allowed without a concomitant loss of analgesic efficacy. However, methylene substitution for NH is not universally tolerated as **11** and **13** are poorly active.

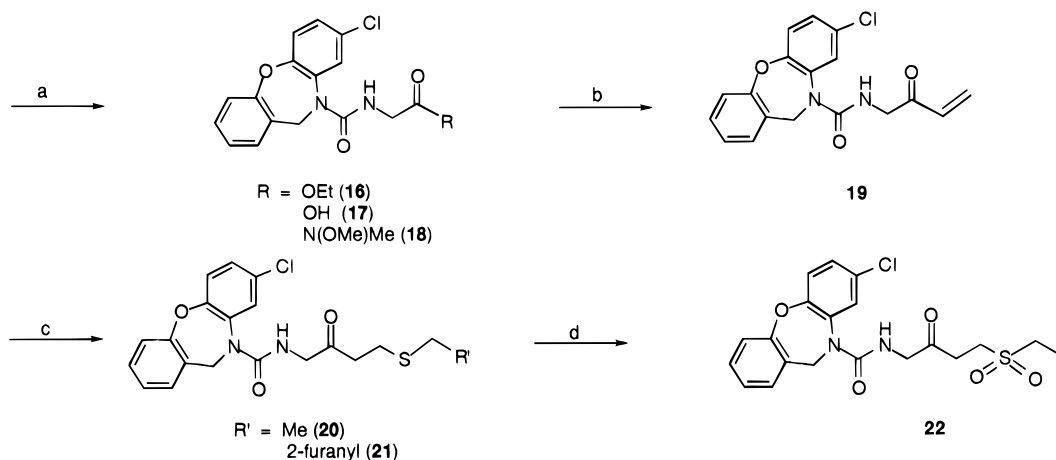
Experimental Section¹

1,3-Dihydro-1,3-dioxo-2H-isoindole-2-acetyl Chloride (6). A solution of 1,3-dihydro-1,3-dioxo-2H-isoindole-2-acetic acid (Phth-Gly) (7.00 g, 34.1 mmol) and thionyl chloride (12 mL, 165 mmol) was heated at reflux for 1 h. Thionyl chloride was distilled from the reaction, and the resulting crude product was further purified by distillation at 160–180 °C (10 Torr) to yield 6.88 g (90%).

8-Chloro-10,11-dihydro-10-[(1,3-dihydro-1,3-dioxo-2H-isoindol-2-yl)acetyl]dibenz[b,f][1,4]oxazepine (7). After heating 3 h at reflux, the reaction mixture of **5** (4.00 g, 17.9 mmol), **6** (4.16 g, 18.0 mmol), and triethylamine (TEA) (2.8 mL, 20.1 mmol) in toluene solution (100 mL) was poured onto 1 N HCl at 0 °C. The insoluble product was collected by filtration, washed with toluene and Et₂O, and dried *in vacuo* to yield 6.36 g (85%) of shiny white crystals: mp 184–185 °C.

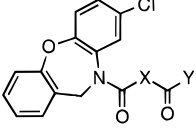
10-(Aminoacetyl)-8-chloro-10,11-dihydrodibenz[b,f][1,4]-oxazepine (8). After a solution of hydrazine monohydrate (1.25 mL) and **7** (5.00 g, 11.9 mmol) in absolute ethanol (125 mL) was heated at reflux for 4 h, the reaction mixture was filtered hot and rinsed with EtOH and then with hot DCM. The filtrate and rinses were combined, evaporated *in vacuo*, and purified by chromatography to give 2.91 g (84%) of a light yellow solid: mp 110–112 °C.

3-(4-Pyridyl)propanoic Acid (9). After stirring for 3 h at ambient temperature, the reaction mixture of methyl 3-(4-pyridyl)propanoate¹ (1.00 g, 6.1 mmol) and sodium hydroxide

Scheme 3^a

^a (a) (i) **5**, OCNCH₂COOEt, toluene, Δ; (ii) 1 N NaOH, MeOH; (iii) 1,1'-carbonyldiimidazole, NH(OMe)Me·HCl, TEA, THF, 5'–20 °C; (b) CH₂=CHMgBr, THF, –70 °C; (c) HSR, piperidine, BzI Me₃N⁺OH[–], DCM/MeOH (4:1); (d) 30% H₂O₂, HOAC, 60 °C.

Table 1. Phenylbenzoquinone Writhing Assay in Mouse and PGE₂ Antagonism in Guinea Ileum for Aminoacetyl Analogs

								
no.	X	ED ₅₀ (mg/kg) ^a	pA ₂ ^b	no.	X	Y	ED ₅₀ (mg/kg)	pA ₂
1	NHNH	6.8 ^c	6.5 ^c	11	CH ₂ NH	(CH ₂) ₂ -4-pyridyl	5/10	5.7 ± 0.05
2	NHNH	9.1 ^c	7.5 ^c	12	CH ₂ NH	CF ₂ CHOH-2-pyridyl	8.1 (0.88–29.6)	nt ^f
3	NHNH	0.9 ^d	8.1 ^d	13	CH ₂ NH	(CH ₂) ₂ SCH ₂ -2-furanyl	13.5 (4.7–199)	7.4 ± 0.13
				21	NHCH ₂	(CH ₂) ₂ SCH ₂ -2-furanyl	5/10	7.2 ± 0.10
4	NHNH	7.4 ^e	5.8 ^e	15	CH ₂ NH	(CH ₂) ₂ SO ₂ CH ₂ CH ₃	5.9 (3.9–16.6)	5.7 ± 0.12
				22	NHCH ₂	(CH ₂) ₂ SO ₂ CH ₂ CH ₃	5/10	5.6 ± 0.10

^a The initial screening dose of test compound is 30 mg/kg. ED₅₀'s are determined for compounds in which seven out of ten animals respond. Values in parentheses are confidence limits determined at 95% ($p < 0.05$). ^b pA₂'s are determined based on the dose ratio at 3 μM.^{9a,13} ^c Reference 1. ^d Reference 10. ^e Reference 11. ^f Not tested.

(0.42 g, 10.5 mmol) in 8 mL of methanol–H₂O (5:3) was treated with 6 N HCl. After the reaction mixture was concentrated, the residue was treated with absolute EtOH (10 mL), heated at reflux, filtered, and rinsed with hot EtOH. The filtrates were evaporated *in vacuo*. The residue was treated with Et₂O, and the product was collected by filtration, washed with Et₂O, and dried *in vacuo* to yield 0.52 g (46%) of a white solid: mp 204–207 °C.

α,α-Difluoro-β-hydroxy-3-(2-pyridyl)propanoic Acid (10). **10** was synthesized in a similar manner to **9** on a 17 mmol scale starting with ethyl α,α-difluoro-β-hydroxy-3-(2-pyridinyl)propanoate¹ to give a quantitative yield of a pale yellow solid.

N-[2-(8-Chloro-10,11-dihydrodibenz[*b,f*]1,4]oxazepin-10-yl)-2-oxoethyl]-4-pyridinepropanamide, Monohydrochloride (11). To a stirring solution of **8** (0.77 g, 2.7 mmol), **9** (0.50 g, 2.7 mmol), and TEA (0.8 mL, 5.8 mmol) in 10 mL of dimethylformamide (DMF) at 5 °C was added 1-(3-(dimethylamino)propyl)-3-ethylcarbodiimide hydrochloride (0.56 g, 2.9 mmol). After stirring at room temperature for 60 h, the reaction was partitioned between saturated NaHCO₃ solution (100 mL) and EtOAc (75 mL) and filtered. The filtrate was separated, and the aqueous layer was extracted with EtOAc (2 × 50 mL). The combined organic extracts were washed with saturated NaHCO₃ solution (2 × 100 mL), H₂O (100 mL), and brine (100 mL) and worked up in the routine manner. The crude product was purified by chromatography. The purified product was taken up in EtOAc and treated with HCl in ethanol. The product was collected by filtration and dried *in vacuo* to yield 0.23 g (19%) of a white solid: ¹H NMR (DMSO-*d*₆) δ 8.74 (d, 2H, $J = 6.6$ Hz), 8.25 (t, 1H, $J = 5.4$ Hz), 7.84 (d, 2H, $J = 6.6$ Hz), 7.49–7.51 (m, 1H), 7.41 (d, 1H, $J = 8.5$ Hz), 7.17–7.30 (m, 3H), 7.11 (dt, 1H, $J = 0.9, 7.7$ Hz), 3.03 (t, 2H,

$J = 7.3$ Hz), 2.56 (t, 2H, $J = 7.3$ Hz). Anal. (C₂₃H₂₀ClN₃O₃·HCl) C, H, N, Cl.

N-[2-(8-Chlorodibenz[*b,f*]1,4]oxazepin-10(11*H*)-yl)-2-oxoethyl]-α,α-difluoro-β-hydroxypyridine-2-propanamide, Monohydrochloride (12). **12** was synthesized in a manner similar to **11** on a 3.8 mmol scale starting with **8** and **10** to give 1.40 g (78%) of the free base which was treated with 1 N HCl to give a white solid hydrochloride salt: ¹H NMR (DMSO-*d*₆) δ 8.69 (d, 1H, $J = 4.6$ Hz), 8.29 (dt, 1H, $J = 1.6, 7.9$ Hz), 7.87 (d, 1H, $J = 8.0$ Hz), 7.76–7.80 (m, 2H), 7.52 (br d, 1H), 7.44 (br d, 1H), 7.27–7.31 (m, 2H), 7.21 (d, 1H, $J = 1.1$ Hz), 7.19 (d, 1H, $J = 1.1$ Hz), 7.11 (t, 1H, $J = 7.3$ Hz), 5.43 (dd, 1H, $J = 8.0, 15.2$ Hz). Anal. (C₂₃H₁₈ClN₃O₃F₂·HCl) C, H, N, Cl.

N-[2-(8-Chloro-10,11-dihydrodibenz[*b,f*]1,4]oxazepin-10-yl)-2-oxoethyl]-3-[(2-furanylmethyl)thio]propanamide (13). Trimethylaluminum (2.0 M in toluene; 4.6 mL) was added dropwise to a stirring solution of methyl 3-[(2-furanylmethyl)thio]propanoate (0.75 g, 3.7 mmol) and **8** (1.00 g, 3.5 mmol) in toluene (30 mL) at room temperature. After stirring for 5 h, the reaction was quenched with methanol (14 mL). After concentrating *in vacuo*, the resulting orange residue was partitioned between chloroform (50 mL) and 1 N NaOH (50 mL). The aqueous layer was extracted with chloroform (50 mL). The combined chloroform extracts were washed with 1 N NaOH (50 mL), 1 N HCl (2 × 50 mL), and brine (50 mL) and worked up in the usual manner. The crude product was purified by chromatography followed by recrystallization from cyclohexane/EtOAc to yield 0.63 g (40%) of a tan solid: mp 90–95 °C. Anal. (C₂₃H₂₁ClN₂O₄S) C, H, N, Cl, S.

N-[2-(8-Chloro-10,11-dihydrodibenz[*b,f*]1,4]oxazepin-10-yl)-2-oxoethyl]-3-(ethylthio)propanamide (14). To a stirring solution of **8** (1.50 g, 5.2 mmol) and triethylamine (0.84

mL, 5.7 mmol) in DCM (40 mL) at 5 °C was added dropwise 3-(ethylthio)propanoyl chloride (0.87 g, 5.9 mmol) in DCM (10 mL). After stirring for 3 h at ambient temperature, the reaction was diluted with DCM (50 mL) and washed with 50 mL each of 1 N HCl, saturated NaHCO₃ solution, and brine. The organic layer was worked up in the usual manner to yield 1.90 g (90%): mp 148–151 °C.

N-[2-(8-Chloro-10,11-dihydrodibenz[*b,f*][1,4]oxazepine-10-yl)-2-oxoethyl]-3-(ethylsulfonyl)propanamide (15). To a stirring solution of **14** (1.50 g, 3.7 mmol) in acetic acid (11 mL) at 58 °C was added dropwise H₂O₂ (30 wt % in water; 1.19 mL). After 1 h, the temperature of the oil bath was raised to 72 °C, and additional H₂O₂ (0.8 mL) was added. After 3 h, the reaction was concentrated. The residue was taken up in EtOAc (50 mL) and washed with saturated NaHCO₃ solution. The aqueous wash was extracted with EtOAc (50 mL), and the organic extracts were combined and washed with 50 mL each of saturated NaHCO₃ solution. The organic layers were worked up in the routine manner and chromatographed. The product was recrystallized from ethanol (3A) to yield 1.30 g (80%) of shiny white crystals: ¹H NMR (DMSO-*d*₆) δ 8.36 (t, 1H, *J* = 5.7 Hz), 7.83 (br s, 1H), 7.49–7.51 (m, 1H), 7.41 (d, 1H, *J* = 8.7 Hz), 7.26–7.29 (m, 2H), 7.19 (d, 1H, *J* = 7.4 Hz), 7.09 (t, 1H, *J* = 7.4 Hz), 3.21 (t, 2H, *J* = 7.5 Hz), 3.05 (q, 2H, *J* = 7.3 Hz), 2.56 (t, 2H, *J* = 7.5 Hz), 1.17 (t, 2H, *J* = 7.5 Hz); mp 167–168 °C. Anal. (C₂₀H₂₁ClN₂O₅S) C, H, N, Cl, S.

Ethyl [(8-Chloro-10,11-dihydrodibenz[*b,f*][1,4]oxazepine-10-yl)carbonyl]amino]acetate (16). After **5** (5.00 g, 21.6 mmol) and ethyl isocyanacetate (3.07 g, 27.1 mmol) in 50 mL of toluene were heated at reflux for 4 h, additional ethyl isocyanacetate (1.00 g, 7.7 mmol) was added. The reaction mixture was heated for 16 h. The reaction was worked up in the usual manner to give 7.12 g (91%) of a yellowish oil which solidified on standing.

[8-Chloro-10,11-dihydrodibenz[*b,f*][1,4]oxazepine-10-yl)carbonyl]amino]acetic Acid (17). After stirring for 16 h, a solution of **16** (6.77 g, 18.8 mmol) in methanol (350 mL) and 1 N NaOH (25 mL) was concentrated. The residue taken up in water (300 mL). The solution was washed with Et₂O (2 × 300 mL), acidified with 1 N HCl, and extracted with Et₂O (2 × 350 mL). The combined ethereal extracts were worked up in the usual manner to yield 6.29 g (100%).

2-[(8-Chloro-10,11-dihydrodibenz[*b,f*][1,4]oxazepine-10-yl)carbonyl]amino]-N-methoxy-N-methylacetamide (18). To a stirring solution of 1,1'-carbonyldiimidazole (2.12 g, 14.3 mmol) in THF (60 mL) at 5 °C was added dropwise **17** (4.00 g, 13.7 mmol) in THF (20 mL). After stirring for 1 h at 5 °C and 4 h at room temperature, TEA (1.9 mL, 13.7 mmol) and *N,O*-dimethylhydroxylamine hydrochloride (1.30 g, 13.3 mmol) were added. After 48 h, the reaction was concentrated and worked up in the usual manner to yield 3.20 g (71%).

8-Chloro-N-(2-oxo-3-butenyl)dibenz[*b,f*][1,4]oxazepine-10(11*H*)-carboxamide (19). To a stirring solution of **18** (3.2 g, 8.5 mmol) in THF (35 mL) at –70 °C was added dropwise a solution of vinylmagnesium bromide (1.0 M THF, 28 mL) in 15 mL of THF. The ice bath was removed, and the reaction mixture was stirred to room temperature for 1 h. The reaction was worked up in the usual manner to yield 2.25 g (77%) of a colorless glass.

8-Chloro-N-[4-(ethylthio)-2-oxobutyl]dibenz[*b,f*][1,4]oxazepine-10(11*H*)-carboxamide (20). A solution of **19** (1.00 g, 2.9 mmol), ethanethiol (0.23 mL, 3.1 mmol), piperidine (0.092 mL), and *N*-benzyltrimethylammonium hydroxide (0.092 mL) in 20 mL of DCM:MeOH (4:1) was stirred for 5 h. The reaction was worked up in the usual manner to yield 0.38 g (32%) of a colorless oil.

8-Chloro-N-[4-(ethylsulfonyl)-2-oxobutyl]dibenz[*b,f*][1,4]oxazepine-10(11*H*)-carboxamide (22). To a stirring solution of **20** (0.38 g, 0.9 mmol) in HOAc (3 mL) at 60 °C was added 30% H₂O₂ (0.3 mL). After 1 h, an additional 0.1 mL of H₂O₂ was added, and the reaction was heated for additional 1 h. The reaction mixture was concentrated and worked up in the usual manner to give 0.16 g (39%) of a white solid: ¹H NMR (DMSO-*d*₆) δ 7.47 (d, 1H, *J* = 2.5 Hz), 7.15–7.42 (m, 5H), 7.06 (dt, 1H, *J* = 1.1, 7.4 Hz), 6.79 (t, 1H, *J* = 5.4 Hz), 4.84 (s, 2H), 3.83 (d, 2H, *J* = 5.5 Hz), 3.23 (t, 2H, *J* = 7.5 Hz),

3.07 (q, 2H, *J* = 7.5 Hz), 2.84 (t, 2H, *J* = 7.4 Hz); mp 140–142 °C. Anal. (C₂₀H₂₁ClN₂O₅S) C, H, N, S.

8-Chloro-N-[4-[(2-furanylmethyl)thio]-2-oxobutyl]dibenz[*b,f*][1,4]oxazepine-10(11*H*)-carboxamide (21). Conditions described for **20** repeated on a 2.2 mmol scale to yield **21**, a colorless oil: ¹H NMR (DMSO-*d*₆) δ 7.57 (dd, 1H, *J* = 0.9, 1.8 Hz), 7.47 (d, 1H, *J* = 2.4 Hz), 7.32–7.41 (m, 2H), 7.15–7.27 (m, 3H), 7.05 (dt, 1H, *J* = 1.3, 7.4 Hz), 6.72 (t, 1H, *J* = 5.6), 6.37 (dd, 1H, *J* = 1.9, 3.2 Hz), 6.24 (dd, 1H, *J* = 0.8, 3.2 Hz), 4.83 (s, 2H), 3.76 (d, 2H, *J* = 5.5 Hz), 3.74 (s, 2H), 2.50–2.64 (m, 4H). Anal. (C₂₃H₂₁ClN₂O₄S) H, N, Cl, S; C: calcd, 60.46; found, 59.94.

Mouse Writhing Assay.¹ The phenylbenzoquinone (PBQ) writhing assay in mouse was used to evaluate the antinociceptive activity.¹

PGE₂ Antagonism Assay Utilizing the Guinea Pig Ileum.¹ PGE₂ antagonism was assessed in PGE₂-stimulated guinea pig ileum muscle strip assay.¹

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References

- Hallinan, E. A.; Hagen, T. J.; Husa, R. K.; Tsymbalov, S.; Rao, S. N.; vanHoeck, J.-P.; Rafferty, M. F.; Stapelfeld, A.; Savage, M. A.; Reichman, M. N-Substituted Dibenzoxazepines as Analgesic PGE₂ Antagonists. *J. Med. Chem.* **1993**, *36*, 3293–3299.
- Malmberg, A. B.; Rafferty, M. F.; Yaksh, T. L. Antinociceptive Effect of Spinally Delivered Prostaglandin E Receptor Antagonists in the Formalin Test on the Rat. *Neurosci. Lett.* **1994**, *173*, 193–196.
- Lee, K.; Vandenbergh, Y. L.; Herin, M. F.; Cavalier, R.; Beck, D.; Li, A.; Verbeke, N.; Lesne, M. L.; Roba, J. Comparative Metabolism of SC-42867 and SC-51089, Two PGE₂ Antagonists, in Rat and Human Hepatocyte Cultures. *Xenobiotica* **1994**, *24*, 25–36.
- IARC Monographs **1974**, *4*, 127–136.
- (a) Taiwo, Y. O.; Levine, J. D. Characterization of the Arachidonic Acid Metabolites Mediating Bradykinin and Noradrenaline Hyperalgesia. *Brain Res.* **1988**, *458*, 402–406. (b) Taiwo, Y. O.; Goetzl, E. J.; Levine, J. D. Hyperalgesia Onset Latency Suggests a Hierarchy of Action. *Brain Res.* **1987**, *423*, 333–337. (c) Lamotte, R. H.; Thalhammer, J. G.; Torebjork, H. E.; Robinson, C. J. Peripheral Neural Mechanisms of Cutaneous Hyperalgesia Following Mild Injury by Heat. *J. Neurosci.* **1982**, *2*, 765–782. (d) Lembeck, F.; Popper, H.; Juan, H. Release of Prostaglandins by Bradykinin as an Intrinsic Mechanism of its Algesic Effect. *Naunyn. Arch. Pharmacol.* **1976**, *294*, 69–73.
- (a) Vane, J. R. Inhibition of Prostaglandin Synthesis as a Mechanism of Action for Aspirin-like Drugs. *Nature* **1971**, *231*, 232–235. (b) Ferreira, S. H. Prostaglandins, Aspirin-like Drugs, and Analgesia. *Nature* **1972**, *240*, 200–203. (c) Ferreira, S. H.; Moncada, S.; Vane, J. R. Prostaglandins and the Mechanism of Analgesia Produced by Aspirin-like Drugs. *Br. J. Pharmacol.* **1973**, *49*, 86–97.
- (a) Abramson, S. B.; Weissmann, G. The Mechanisms of Action of Nonsteroidal Antiinflammatory Drugs. *Arthritis Rheumatism* **1989**, *32*, 1–9. (b) Gelgor, L.; Butkow, N.; Mitchell, D. Effects of Systemic Non-steroidal Anti-inflammatory Drugs on Nociception During Tail Ischaemia and on Reperfusion Hyperalgesia in Rats. *Br. J. Pharmacol.* **1992**, *105*, 412–416.
- (a) Isselbacher, K. J. The Role of Arachidonic Acid Metabolites in Gastrointestinal Homeostasis. *Drugs* **1987**, *33* (Suppl. 1), 38–46. (b) Whittle, B. J. R. *ISI Atlas Sci.: Pharmacol.* **1987**, 168–172. (c) Miller, T. A. Protective Effects of Prostaglandins Against Gastric Mucosal Damage. *Am. J. Physiol.* **1983**, *245*, G601–23. (d) Konturek, S. J. Gastric Cytoprotection. *Scand. J. Gastroenterol.* **1985**, *20*, 543–553. (e) Cohen, M. M., Ed. *Biological Protection with Prostaglandins*; CRC Press: Boca Raton, FL, 1986; Vol. II.

- (9) (a) Sanner, J. H. Antagonism of Prostaglandin E_2 by 1-Acetyl-2(8-chloro-10,11-dihydrodiben[b,f]oxazepine-10-carbonyl hydrazine. (b) Drower, E. J.; Stapelfeld, A.; Mueller, R. A.; Hammond, D. L. The Antinociceptive Effects of Prostaglandin Antagonists in the Rat. *Eur. J. Pharmacol.* **1987**, *133*, 249–256. (c) Sanner, J. H.; Mueller, R. A.; Schulze, R. H. Structure-Activity Relationships of Some Dibenzoazepine Derivatives as Prostaglandin Antagonists, *Adv. Biosci.* **1972**, *9*, 139–148. (d) Sanner, J. H. Dibenzoazepine Hydrazides as Prostaglandin Antagonists. *Intra-Science Chem. Rep.* **1972**, *6* (1), 1–9.
- (10) Hallinan, E. A.; Stapelfeld, A.; Savage, M. A.; Reichmann, M. 8-Chlorodibenz[b,f][1,4]oxazepine-10(11H)carboxylic acid, 2-[3-[(2-furanylmethyl)-thio]-1-oxopropyl]hydrazide (SC-51322): A Potent PGE_2 Antagonist and Analgesic. *Bioorg. Med. Chem. Lett.* **1994**, *4*, 509–514.
- (11) Husa, R. K.; Rafferty, M. F.; Hagen, T. J.; Hallinan, E. A. Unpublished data.
- (12) Mueller, R. A.; Hammond, D. L. Manuscript in preparation.
- (13) Nogardy, T. *Medicinal Chemistry: A Biochemical Approach*; Oxford University Press: New York, 1985; pp 60–65 and references cited therein.

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